

### Name of the Department: Department of P G Studies and Research in Social Work

No of papers revised in	Units undergone	Unit wise provide elaborate student
the Syllabus (Provide	revision(Provideentire	specific outcome learning(what student
		learns from revised unit)
		*The content is reshuffled and segregated to
		give an orderly structure to enable the
	5	student to understand and relate the relevant
		concepts pertaining to social work.
Marks:70	ethics , professionalism,	*Basic social work concepts are put together to ensure easy understanding of the
		concepts to the non Social Work
	5	background students
		*Professionalism, attributes of profession,
		requisites for social work profession are
	pertaining to practice of social work is put here which includes	taught in the unit
	models, human rights etc	*Relevant constitutional provisions for the
		welfare of the society is taught.
	6	
		* All the units are revised with the intention
		of making it convenient for students to
		relate the concepts well . few concepts are
		added to give clear detail of what the
		students learn under each heading.
Marks:70		
	,	
	,	
	the Syllabus (Provide details of the paper) Course : Social Work Profession Total Hours:48 Code:SWSH401 Course Type:CBCS Marks:70 Course: Medical Social Work Total Hours:40 Code:SWSS504 Course Type: CBCS Marks:70	details of the paper)revised unit)Course : Social WorkUnit 1: Basic concepts relevant to social work are added.Total Hours:48Introduction and history ofCode:SWSH401Social work Profession is brought to this unit.Marks:70Unit 2: Attributes, values , ethics , professionalism, introduction to core and auxiliary methods of social work , Social Work education is clubbed together in this unit Unit 3: All the concepts pertaining to practice of social work is put here which includes models,human rights etc Unit 4: constitutional guarantees for social welfare is added.Course: Medical Social WorkUnit 1: concept of prevention, community health care, principles of health care is addedCourse Type: CBCSUnit 2: Names of

#### **1.1Curriculum Design And Development**



Course :Social Work	Unit 1:	Students will update about trends and
Practice with	Recent trends in Social case	adopt the skills required.
Individuals and Group	work	Students will learn the Job
Total Hours:40	Unit 2:	Opportunities and will make the
Code:SWS H 402	Role and Job Scope for Case	preparations for the same. Helps in
Course Type: Theory	Worker.	specific focus
Marks:70	Unit 3:	Students will update about trends and
	Recent trends in Social	adopt the skills required.
	group Work.	Students will understand the method of
	Models in group work	implementing strategies.
	Unit 4:	
	Role and Job Scope for	Students will learn the Job
	Group Worker.	Opportunities and will make the
	Study of Case Work and	preparations for the same. Helps in
	Group Work format	specific focus.
	1	Helps in Social Practicum casework
		recording
Course :		Students will orient and learn skills
THERAPEUTIC	Unit 2:	techniques and approaches in online
COUNSELLING	Online Counseling.	counselling which is getting lot of
Total Hours:40		prominence and also a recent trend in
Code: SWS S 555	Unit 4:	counselling.
Course Type: Theory	Case studies, Life skills	Students will get practical orientation
Marks:70	helping model, Role Play,	about counselling methods, application,
	Practical Counseling	target group, techniques etc.
	sessions. Life style	
	modification and wellness	Along with this students will learn job
	counseling Model-	Thong with this students will learn job
	Hospital for nature cure and	
	yoga therapy,	scope
	ShanthiyanaDharmasthala.	
	Addiction counseling –	
	Model- JanaJagruthiVedike,	
	SDM trust –Laila.	
	Counseling women in	
	distress and violence –	
	Model- GelathiCounseling	
	centre SIRI Laila.	
	SANTHWANA. Women's	
	helpline Belthangady.	
	PrajnaCounseling centre	
	MangaloreBehaviour	
	therapy- Model-Anirveda	
	Mangalore.	
	Palliative Care- Model- Ave	
	Maria, Vamanjooru	



Course :REHABILITATION AND AFTER CARE SERVICES Total Hours:40 Code: SWS S 557-A Course Type:Theory Marks:100	Unit 4: Orientation Visits to Rehabilitation Settings. Study of Various Rehabilitation Models Ministry of Empowerment and social Justice (Dept of welfare of disabled). Role of Social Worker in District Rehabilitation centre. National Rehabilitation centers.	Students will orient and learn skills techniques and approaches in online counselling which is getting lot of prominence and also a recent trend in counselling. Students will get practical orientation about Rehabilitation models, methods, application, target group, techniques etc. Along with this students will learn job scope. Students will orient about agencies working under rehabilitation and they will identify the job scope for MSW profession in these agencies. Students will orient about skills required in Rehabilitation sector. Students will orient about agencies at the national level and skills required in Rehabilitation sector.
Course : Non- Governmental Organization and Societal Development Total Hours: 34 Code: SWS 509 B Course Type: Open Elective Paper Marks: 70	Unit 2: Central Social Welfare Board – Objectives, Administrative Structure, Functions and Programs	Understands CSWB, its objectives, administrative structure and functions and programmes
Course : Urban Community Development Total Hours:34 Code: SWSS 551 Course Type: CD Specialization Paper Marks : 70	Unit 3: Models of Urban Development Ministry of Housing and Urban Affairs and Its programmes Karnataka Urban Development and Coastal Environment Management Plan – A case study Unit 4 : - Role of NGOs in Urban	Unit 3: Students will able to learn different models of Urban Development like Sustainable Model of Urban Development, Smart City etc Students will able to learn different programmes and plans of Ministry of Housing and Urban Affairs and its structure and composition Students will able to Learn, Analyse and Evaluate the success and failures and drawbacks, Challenges, limitations and



# Sri Dharmasthala Manjunatheshwara College

(Autonomous), Ujire-574 240, Dakshina Kannada, Karnataka State

	Development Intervention of Corporate Sector for Urban Development (CSR)	opportunities in relation to programmes and plans of Karnataka Urban Development and Coastal Environment Management Unit 4: Students will able to learn how NGOs are contributing to the urban development through its innovative programmes and services Students will able to learn how Corporate sector under its CSR (Corporate Social Responsibility) initiatives and funds plan to contribute for the urban development
Course: Psychiatric Social Work Total Hours:48 Code: SWSS503 Course Type: Specialisation Marks:70	Unit 2: Current trends in Social Work. Limitations faced by the social worker in the field and alternative to improve the professional status of Psychiatric Social Work in India-NIMHANS Model. Role of PSW in Government Hospital, Unit 3: Intellectual Disability Psychiatric Case History Taking Format	Unit 2: Students will understand the changing milieu in Social Work Students will know the limits and challenges of Mental Health Field Students will understand the various frame of reference if they study the models of mental health Students will gain knowledge on the different roles played by psychiatric social worker in mental health field. Unit 3- Used the new term Students will understand the procedures to be followed during the psychiatric case history taking format. Used new term
Course: Social Work Research and Statistics Total Hours:48 Code: SWS453 Course Type: General Paper Marks:70	Unit 1: Assumptions of science. Scientific approach in comparison to the native or the common-sense approach Application of scientific approach in study of social phenomena. Unit-2: Rationale, General consideration in determination of sample size. Scales-need for scales and some prominent scaling procedures. Unit 3: Nil Unit 4: Variance and 'F'distribution, SPSS Package.	Unit-1 Students will gain the opportunity to gain better knowledge of how and why things function Unit-2 Students will understand in- depth knowledge in the field of sampling Students will gain the scientific knowledge on data collection and analysing the data. Unit-4 Students will enhance their scientific knowledge on statistics.



#### Name of the Department: PG Studies and Research in Chemistry

### M.Sc. Chemistry

		<b>TT T T T T T T T T </b>	<b>TT 1 1</b>
Revision carried in 2022	No of papers revised in the syllabus(Provide	Units undergone revision( Provide entire revised unit)	Unit wise provide elaborate student specific outcome
	details of the paper)		learning( what student learns from
			revised unit)
	Course:	Metallurgy and redox	Understanding of
	Inorganic	potentials	distinct type of
	Chemistry, Unit II	Methods of reduction of	solvents
	Total Hours: 15 hrs	oxide ores, chemical and	
	Code: CH H 451	electrolytic reductions,	
	Course Type:	Ellingham diagram,	
	Theory	Specialized techniques for	
	Marks: 100	the extraction of metals – Amalgamation,	
		Hydrometallurgy, Solvent	
		Extraction, Ion exchange	
		chromatography. Reduction	
		potentials, Latimer and	
		Frost diagrams – features	
		and applications.	
		<b>Reactions in non-aqueous</b>	
		media:Types of solvent,	
		Characteristics of solvents.	
		Anhydrous sulphuric acid,	
		glacial acetic acid,	
		anhydrous HF, bromine	
		trifluoride, liquid sulphur	
		dioxide and dinitrogen	
		tetroxide. Reactions in	
		molten salts. Self Study: Liquid	
		Self Study: Liquid ammonia	
	Course:	Reaction Mechanisms in	Applications of
	Coordination	Transition Metal	organometallic
	Chemistry, Unit III	<b>Complexes:</b> Energy	compounds
	Total Hours: 15 hrs	profile of a reaction, inert	L
	Code: CH H 501	and labile complexes,	
	Course Type:	kinetics of octahedral	
	Theory	substitution and	
	Marks: 100	mechanistic aspects. Acid	
		hydrolysis, factors affecting	
		acid hydrolysis, base	
		hydrolysis, conjugate base	
		mechanism and evidences	
		in its favour. Anation	



	reactions, reactions without M-L bond cleavage. Substitution reactions in square planar complexes, trans effect, mechanisms of substitution. Electron transfer reactions- inner sphere and outer sphere reactions, complimentary and non-complimentary reactions. <b>Organometallic</b> <b>compounds:</b> Introduction, Nomenclature: Hapticity, 16/18 electron rule, EAN, Applications of organometallic compounds.	
Course: Chemistry of Synthetic Drugs, Unit I Total Hours: 15 hrs Code: CH S 504 Course Type: Theory Marks: 100	Introduction. Drug design and relationship of functional groups to pharmacologic activity- introduction, relationship between molecular structure and biologic activity, selectivity of drug action and drug receptors, biologic targets for drug action, physicochemical properties of drugs, stereochemistry and drug action drug design- discovery and structural modification of lead compounds, physicochemical and biopharmaceutical properties of drug substances, pharmacokinetics- physicochemical factors affecting drug absorption.Induced fit theory of drug action <b>General anaesthetics:</b> Synthesis of Halothane, & Methohexital sodium. <b>Local anaesthetics:</b> Synthesis and mode of action of Benzocaine & Procaine Hydrochloride,	Promotion to drug discovery research



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Course:	A. Kinetics and Catalysis	Exposure and use
Physical Chemistry	(Any Four Experiments	of advanced
Practicals-IV	are to be carried out)	instruments
Total Hours: 6 hrs	Determination of reaction	
Code: CH P 558	order and activation	
Course Type:	parameters, study of	
Practical	acidity/salt/solvent/catalytic	
Marks: 100	effects on reaction rates of	
Widiks. 100	any FIVE of the reactions	
	listed	
	below.	
	1. Acid catalyzed hydrolysis	
	of methyl acetate.	
	2. Saponification of ethyl	
	acetate by conductivity	
	method.	
	3. Decomposition of	
	benzenediazonium chloride.	
	4. Reaction between	
	potassium persulphate and	
	potassium iodide (including	
	the study of salt	
	effect and catalysis by Ag <sup>+</sup> ,	
	Fe $^{2+}$ and Cu $^{2+}$ ions).	
	5. Decomposition of	
	diacetone alcohol by	
	NaOH& Hydrolysis of t-	
	butylchloride.	
	6. (i) Reaction between	
	iodine and acetone, and (ii)	
	iodination of aniline.	
	7. Reaction between	
	hydrogen peroxide and HI.	
	8. Decomposition of $H_2O_2$	
	(including the study of	
	catalytic effect).	
	9. Reaction between	
	Chromic acid and oxalic	
	acid.	
	10. Reduction of aqueous	
	solution of ferric chloride by	
	stannous chloride.	
	11. Determination of the	
	mechanism of the oxidation	
	of an organic compound	
	from kinetic	
	data.	
	12. Determination of	
	catalytic constant of an acid.	
	13. Determination of effect	
	of surface area of catalyst	
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	and temperature on the	
	kinetics of	
	Metal-acid reaction.	
	14. Determination of	
	dissociation of	
	trichloroacetic acid-Kinetic	
	method.	
	15. Determination of	
	equilibrium constant for	
	homogeneous equilibria and	
	determining the	
	concentration of a given	
	solution.	
	16. Determine the molecular	
	formula of copper-ammonia	
	complex by the partition	
	coefficient method.	
	17. Alkaline hydrolysis of	
	ethyl acetate volumetrically.	
	18. Effect of reaction surface	
	area of catalyst and	
	temperature, concentration	
	on the kinetics	
	of metal-acid	
	B. Polymer Chemistry	
	(Any Two experiments are	
	to be carried out)	
	1. Determination of	
	molecular weight and size	
	parameters of polymers by	
	viscometry.	
	2. Determination of	
	sequences in	
	*	
	polyvinylalcohol by viscometry.	
	3. Determination of	
	molecular weight of a	
	polymer by turbidimetry.	
	4. Preparation of	
	*	
	Polymethylmethacrylate by	
	suspension polymerization /	
	polystyrene by	
	free radical polymerization /	
	Nylon by interfacial	
	polymerization /	
	Polyacrylamide by	
	solution polymerisation	
	method / polyvinylalcohol	
	from polyvinylacetate /	
	Phenol formaldehyde/ urea	



	formaldehyde resins / thin	
	films of polymers.	
	C. Thermodynamics	
	<b>Experiments (Any Four</b>	
	experiments to be carried	
	out)	
	1. Determination of	
	activities of an electrolyte	
	and non-electrolyte by	
	cryoscopy.	
	2. Determination of partial	
	molar volumes of (a) Salts-	
	water and (b) alcohol-water	
	(methanol & ethanol)	
	systems by density method.	
	3. Study of complex	
	formation between mercury	
	and potassium halides by	
	cryoscopy.	
	4. Determination of specific	
	heat of liquids and solutions	
	by calorimetry. 5. Determination of stepwise	
	neutralisation of acids.	
	6. Determination of heat of	
	solution of KNO <sub>3</sub> in water,	
	integral heat of dilution of	
	H <sub>2</sub> SO <sub>4</sub>	
	and heat of ionization of	
	acetic acid and ammonium	
	hydroxide calorimetrically.	
	7. Cryoscopic and	
	ebullioscopic analysis of the	
	given mixture of urea and	
	glucose.	
	8. Determination of vant	
	Hoff's factor for benzoic and	
	acetic acid mixtures in	
	benzene.	
	9. Velocity of sound in	
	liquid-ultrasonic	
	interferometry	
	10. Corrosion behavior of	
	different additives for mild	
	steel and zinc in acid media	
	using electrochemical	
	methods	
	D. Spectroscopic	
	Experiments (Any Two	
	experiments to be carried	
	out)	



1. Kinetics of oxidation of	
alcohol by potassium	
dichromate –	
spectrophotometrically.	
2.Simultaneous	
determination of Manganese	
and chromium in a solution	
of dichromate and	
Permanganate mixture.	
3. Determination of pKa of	
an indicator.	
4. Spectroscopic	
investigation of partition	
coefficient	
ofiodinebetweenH2O	
andCHCl <sub>3</sub> .	
5. Study of the effect of	
ionic strength on the pH of	
the given acid with the help	
of Indicators using buffer	
solution by colorimetric	
method.	
6. Spectrophotometric	
determination of Critical	
micelle concentration	
E. Radiochemistry	
Experiments (At least Two	
experiments to be carried	
1. Study of (a) Characteristic	
plateau, (b) Geometry	
effects and Statistics of G.M	
counter	
2. Determination of (a) Dead	
time by single source &	
double source method. (b)	
Emax of □-	
source (c) Back scattering of	
$\Box$ and (d) $\Box$ energy emitted	
by C-14.	
3. Verification of the inverse	
square law.	
4. Determination of half life	
of radionuclides.	
4. Determination of Linear	
and mass attenuation	
coefficient.	
5. Preparation of Fricke and	
Ceric sulphate dosimeters &	
calculation of G-value &	
dose rate.	



	<ul> <li>6. Study of isotope dilution analysis;</li> <li>8. Radiochemical Determination of I-131 in sea water.</li> <li>7. Determination of β- particle range and, axmum energy (by half thickness method).</li> <li>8. Percentage purity of copper sulphate by electrogravimetric method.</li> <li>F. Adsorption <ol> <li>Synthesis of a suitable adsorbent (e.g activated</li> </ol> </li> </ul>	
	<ul> <li>and divalent metal ions and their mixture on a suitable adsorbent. Applicability of Freundlich and Langmuir Adsorption isotherms.</li> <li>3. Adsorption characteristics of pollutants such as dyes and/or surfactants on a suitable adsorbent.</li> <li>4. Investigation of adsorption charactrisitics of different dyes (cationic andanionic) on two different types of activated carbons.</li> </ul>	



## M.Sc. Organic Chemistry

Revision carried in	No of papers	Units undergone revision	Unit wise provide
2022	No of papers revised in the	(Provide entire revised	elaborate student
2022	syllabus(Provide	unit)	specific outcome
	details of the paper)	unit)	learning( what
	details of the paper)		student learns from
	Comment		revised unit)
	Course:	Metallurgy and redox	Understand to
	Inorganic	potentials	distinct the type
	Chemistry, Unit II	Methods of reduction of	solvents
	Total Hours: 15 hrs Code: OC H 451	oxide ores, chemical and	
		electrolytic reductions,	
	Course Type:	Ellingham diagram,	
	Theory Marks: 100	Specialized techniques for	
	IVIAIKS. 100	the extraction of metals –	
		Amalgamation, Hydrometallurgy, Solvent	
		Extraction, Ion exchange	
		chromatography.	
		Reduction potentials,	
		Latimer and Frost	
		diagrams – features and	
		applications.	
		<b>Reactions in non-</b> aqueous media:Types of	
		solvent, Characteristics of	
		solvents. Anhydrous	
		sulphuric acid, glacial	
		acetic acid, anhydrous HF,	
		bromine trifluoride, liquid	
		sulphur dioxide and	
		dinitrogen tetroxide.	
		Reactions in molten salts.	
		Self Study: Liquid	
		ammonia	
	Course:	Catalysis by	Understanding the
	Organometallic	Organometallic	applications of
	Chemistry, Unit I	<b>Compounds:</b> Catalysis by	Wacker process
	Total Hours: 15 hrs	Organometallic	×
	Code: OC H 503	Compounds: 16 and 18-	
	Course Type:	electron rules, oxidative	
	Theory	addition, insertion,	
	Marks: 100	deinsertion and reductive	
		elimination reactions.	
		Homogeneous catalysis by	
		organometallics-	
		hydrogenation,	
		hydrosilation,	





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Course:	Synthesis of one	Exposure and use
Organic Cher	nistry derivative each of Furan,	of advanced
Practicals-IV	Indole, Pyrazole,	instruments
Total Hours:	6 hrs Quinoline, Thiazole,	
Code: OC P 5	510 Acridine, Coumarin and	
Course Type:	Triazole containing	
Practical	heterocycles. Synthesis of	
Marks: 100	Picric acid, Para red,	
	Methyl red, Methyl	
	orange, Flourescein,	
	Eosin, Indigo.	
	Chromatographic	
	1	
	column chromatography	
	Elucidation of structure of	
	organic compounds using	
	UV, IR, NMR and Mass	
	spectra.	
	Purification of organic	
	compounds using flash	
	chromatography	
Course:	Isolation and	Exposure and use
Organic Cher	nistry Characterization of natural	of advanced
Practicals-VI	I products like Caffeine,	instruments
Total Hours:	6 hrs Recinoleic acid, Azelic	
Code: OC P 5	559 acid, Piperine,	
Course Type:	Hespiridine, Cysteine,	
Practical	Casein, Lycopene and	
Marks: 100	enzymes like Lipase and	
	Sucrase. Extraction of	
	Groundnut oil and	
	Coconut oil.	
	Determination of	
	Saponification and Iodine	
	values of oils and fats.	
	Isolation of Carotines.	
	Purification by paper, Column and TLC.	
	Column and TLC. Characterization of natural	
	products by oxidation	
	studies & derivatization of	
	natural products.Isolation	
	of carotenoids and	
	purification by Flash	
	chromatography	



Name of the Department: Department of P.G Studies & Research in Psychology				
Revision carried in 2022	No of papers revised in the syllabus (Provide details of the paper)	Units undergone revision (Provide entire revised unit)	Unit wise provide elaborate student specific outcome learning( what student learns from revised unit)	
	Course :Counselling skillsPYH452: counselling skills – content of unit IV is merged with Unit III and the content related to areas of counselling is added as IV Unit.Code: PYH453IV Unit.	Become acquainted with the roles, functions and qualities of an effective counsellor.		
	Course Type: Hard core Marks:70	Title of the paper PYH 452: Counselling skills was changed to Counselling skills and areas of Counselling.	Understand various stages involved in the process of counselling.	
	Course: Psychotherapeutic Interventions Total Hours:40 Code: PYH504 Course Type: Soft core Marks:70	To include Trauma focused therapy and Family therapy in PYH504: Psychotherapeutic interventions.	Analyse the method, merits, limitations and applicability of different psychotherapies.	
	Course: IV SEMESTER Total Hours:48 Code: PYH552 and PYH554 Course Type: Hard core Marks:70	<ul> <li>PYH552: Internship and PYS554 case analysis.</li> <li>PYH552: Areas of Counselling and PYH554: Areas of Counselling practical were removed from IV</li> <li>Semester and the major content were added as IV Unit of PYH 452: Counselling skills.</li> <li>Papers on internship and case were added as PYS552: Internship and PYH554: Case studies and field</li> <li>visit in IV Semester instead of the existing PYH552 Areas of counselling and PYS554: Areas of Counselling Practical.</li> </ul>	To enhance the practical exposure i.e. employability skills.	



a contract of the test of the test of	Course: Marketing, advertising and consumer behaviour Fotal Hours:40 Code: PYS556	Unit I: Concepts related to customer relationship was removed and only concepts related to Marketing is retained.	To provide conceptual knowledge on basic concepts of advertising in order to develop students skills on analysing
C	Course Type: Softcore Marks:70	Unit IV: Foundation of advertisement and effective advertisement is removed as the concepts are Not beneficial for the students of Psychology. Instead the concepts on consumer behavior is shifted From II Unit to Unit IV. Title of the Paper PYS556 has been changed as Marketing and consumer behaviour.	advertisements psychologically and developing effective advertisements.



Name of the Department: Department of PG Studies & Research in Biotechnology			
Revision carried in 2022	No of papers revised in the syllabus (Provide details of the paper)	Units undergone revision( Provide entire revised unit)	Unit wise provide elaborate student specific outcome learning (what student learns from revised unit)
	Course : - Genetic Engineering Unit 1 Total Hours:13 Code: BTH452 Course Type: Theory Marks:	General introduction to concepts of genetic engineering. Host controlled restriction and modification, restriction endonucleases, target sites sticky, cohesive ends and blunt ended fragments. Role of DNA ligase, linkers, adaptors, homopolymer tailing. Other methods of joining DNA molecules: TA cloning of PCR products, Construction of genomic libraries, construction of cDNA library, methods of cDNA synthesis; PCR: Optimization of PCR, RT- PCR and Real time PCR Application of PCR in cloning, agriculture and medicine	PCR different types & its application
	Course:Immunology Total Hours:13 Code:BTH551	Immunoglobulins: Isolation and purification of immunoglobulins. Structure of	Antibody structure & production
	Course Type:Theory	antibodies. Classes	

## Name of the Department: Department of PG Studies & Research in Biotechnology



Marks:	and subclasses of immunoglobulins, biological and chemical properties of Igs. Hyper variable region, isotopic, allotypic and idiotypic variations and idiotypic network. Biosynthesis, theories of formation, diversity of antibodies, genetics of Ig diversity, mechanisms contributing to antibody diversity, Ig genes, isotype switching, Ag-Ab reactions, specificity, affinity binding of antibodies. Production of polyclonal and monoclonal antibodies.	

